

Real gas: Virial expansion

1.

- The ideal gas law ($pV = Nk_B T$) is exact in the limit of no interactions between atoms. A "good approximation" when the density of atoms ($\frac{N}{V}$) is small.
- Corrections to ideal gas law are often expressed in terms of a density expansion, known as the "virial expansion".
- In general,

$$\frac{p}{k_B T} = \frac{N}{V} + B_2(T) \frac{N^2}{V^2} + B_3(T) \frac{N^3}{V^3} + \dots$$

where, $B_j(T)$ are known as the virial coefficients.

- Objective: To compute virial coefficient from first principles. i.e., starting from the knowledge of the underlying potential energy $U(r_{ij})$ between two neutral atoms i and j separated by a distance r_{ij} .

$$\mathcal{H} = \sum_{i=1}^N \frac{|\vec{p}_i|^2}{2m} + \sum_{i < j} U(r_{ij}).$$

Notation:- $\begin{cases} \vec{p}_i = \text{momentum of } i^{\text{th}} \text{ atom.} \\ p = \text{pressure.} \end{cases}$

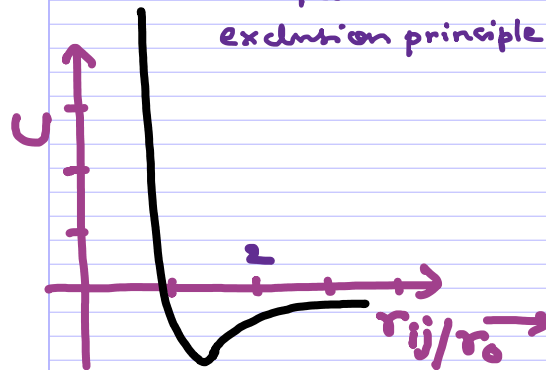
Typical interaction potentials

• Lennard-Jones

$$U(r_{ij}) \sim \left(\frac{r_0}{r_{ij}}\right)^{12} - \left(\frac{r_0}{r_{ij}}\right)^6$$

Pauli's
exclusion principle

due to fluctuating
dipoles of the neutral
atoms.

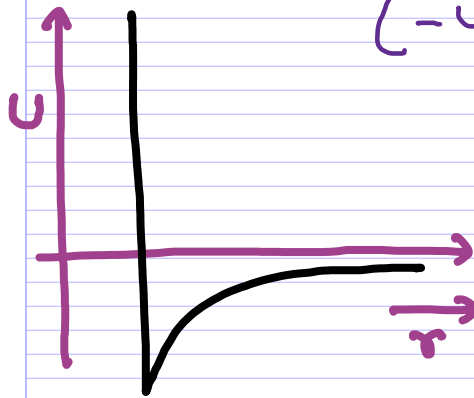


temporary dipole moments

quantum fluctuations

• Hard core

$$U(r_{ij}) = \begin{cases} \infty & r < r_0 \\ -U_0 \left(\frac{r_0}{r}\right)^6 & r \geq r_0 \end{cases}$$



$$Z(N, V, T) = \frac{1}{N!} \frac{1}{(2\pi\hbar)^{3N}} \int \prod_{i=1}^N d^3p_i d^3q_i e^{-\beta H}$$

$$\text{with } H = \sum_{i=1}^N \frac{|\vec{p}_i|^2}{2m} + \sum_{i < j} U(r_{ij}).$$

$$\Rightarrow Z(N, V, T) = \frac{1}{N!} \frac{1}{(2\pi\hbar)^{3N}} \left[\int \prod_{i=1}^N d^3p_i e^{-\beta \sum_{j=1}^N \frac{|\vec{p}_j|^2}{2m}} \right] \\ \times \left[\int \prod_{i=1}^N d^3r_i e^{-\beta \sum_{j < k} U(r_{jk})} \right].$$

(here, $q_i = r_i \neq i$)

$$= \frac{1}{N!} \frac{1}{\lambda^{3N}} \int \prod_{i=1}^N d^3r_i e^{-\beta \sum_{j < k} U(r_{jk})}$$

$$\text{where, } \lambda = \sqrt{\frac{2\pi\hbar^2}{mk_B T}}$$

(thermal de Broglie wavelength)

A short primer on Mayer function

$$\text{Mayer function, } f(r) = e^{-\beta U(r)} - 1.$$

such that

$$f(r) \rightarrow 0 \quad \text{as } r \rightarrow \infty.$$

$$\& f(r) \rightarrow -1 \quad \text{as } r \rightarrow 0.$$

$$\text{Define } f(r_{ij}) = f_{ij}$$

$$\therefore Z(N, V, T) = \frac{1}{N!} \frac{1}{\lambda^{3N}} \int \prod_{i=1}^N d^3r_i \prod_{j < k} (1 + f_{jk})$$

$$\Rightarrow Z(N, V, T) = \frac{1}{N! \lambda^{3N}} \int \prod_{i=1}^N d^3 r_i \left\{ 1 + \sum_{j < k} f_{jk} + \sum_{\substack{j < k \\ l < m}} f_{jk} f_{lm} + \dots \right\}$$

First term in $K \equiv V^N$.

Second term in K has a sum, each element of which is same, such that,

$$\begin{aligned} \int \prod_{i=1}^N d^3 r_i f_{12} &= V^{N-2} \int d^3 r_1 d^3 r_2 f(r_{12}) \\ &= V^{N-1} \int d^3 r f(r). \end{aligned}$$

We make a transformation from

$$\vec{r}_1, \vec{r}_2 \text{ to } \vec{r}, \vec{R}$$

$$\text{where, } \vec{r} = \vec{r}_1 - \vec{r}_2.$$

$$\vec{R} = \frac{\vec{r}_1 + \vec{r}_2}{2}.$$

There are $\frac{N(N-1)}{2}$ such pairs $\approx \frac{N^2}{2}$ pairs
(for $N \rightarrow 10^{23}$.)

$$\therefore Z(N, V, T) = \frac{V^N}{N! \lambda^{3N}} \left\{ 1 + \frac{N^2}{2V} \int d^3 r f(r) + \dots \right\}$$

$$\stackrel{\text{"trick"}}{\approx} \frac{V^N}{N! \lambda^{3N}} \left[1 + \frac{N}{2V} \int d^3 r f(r) \right]^N.$$

$$= Z_{\text{ideal}} \left[1 + \frac{N}{2V} \int d^3 r f(r) \right]^N.$$

$$\ln(1+x) = x - \frac{x^2}{2} + \frac{x^3}{3} - \dots$$

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$$\Rightarrow F = -k_B T \ln Z(N, V, T)$$

$$= F_{\text{ideal}} - N k_B T \ln \left(1 + \frac{N}{2V} \int d^3r f(r) \right)$$

$\approx \frac{N}{2V} \int d^3r f(r)$

$$\therefore p = - \left(\frac{\partial F}{\partial V} \right)_{N, T} = \frac{N k_B T}{V} - \frac{N^2 k_B T}{2V^2} \int d^3r f(r)$$

$$= \frac{N k_B T}{V} \left(1 - \frac{N}{2V} \int d^3r f(r) \right)$$

$$\therefore \frac{pV}{N k_B T} = 1 - \frac{N}{2V} \int d^3r f(r)$$

Note: • $U(r) > 0 \Rightarrow f(r) < 0$
 $\Rightarrow p$ increases.

• $U(r) < 0 \Rightarrow f(r) > 0$
 $\Rightarrow p$ decreases.

Consider a typical potential which is attractive at long distances and repulsive at short distances, e.g., Hard core + vander Waals.

$$\Rightarrow \int d^3r f(r) = \int_0^{r_0} (-1) d^3r + \int_{r_0}^{\infty} d^3r \left(e^{\beta U_0 \left(\frac{r_0}{r} \right)^6} - 1 \right)$$

At high temperatures,

$$e^{\beta U_0 \left(\frac{r_0}{r} \right)^6} \approx 1 + \beta U_0 \left(\frac{r_0}{r} \right)^6$$

$$\Rightarrow \int d^3r f(r) = -4\pi \int_0^{r_0} r^2 dr + 4\pi \beta U_0 \int_{r_0}^{\infty} \frac{r_0^6}{r^4} dr$$

$$= \frac{4\pi r_0^3}{3} \left(\frac{U_0}{k_B T} - 1 \right)$$

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$$\Rightarrow \frac{pV}{Nk_B T} = 1 - \frac{N}{2V} \left\{ \frac{4\pi r_0^3 U_0}{3k_B T} - \frac{4\pi r_0^2}{3} \right\}$$

$$= 1 - \frac{N}{V} \left(\frac{a}{k_B T} - b \right).$$

$$\text{where, } a = \frac{2\pi r_0^3 U_0}{3}; \quad b = \frac{2\pi r_0^2}{3}.$$

$$\therefore \frac{pV}{Nk_B T} = -\frac{Na}{Vk_B T} + \left(1 + \frac{Nb}{V}\right).$$

$$\Rightarrow \frac{V}{Nk_B T} \left(p + \frac{N^2 a}{V^2} \right) = \left(1 + \frac{Nb}{V}\right).$$

$$\therefore k_B T = \frac{V}{N} \left(p + \frac{N^2 a}{V^2} \right) \underbrace{\left(1 + \frac{Nb}{V}\right)^{-1}}_{\approx \left(1 - \frac{Nb}{V}\right)}$$

$$\therefore Nk_B T = \left(p + \frac{N^2 a}{V^2} \right) (V - Nb).$$

⇓

$$\boxed{\left(p + \frac{N^2 a}{V^2} \right) (V - Nb) = Nk_B T}$$

Equation of state for vander Waals gas.

Valid at

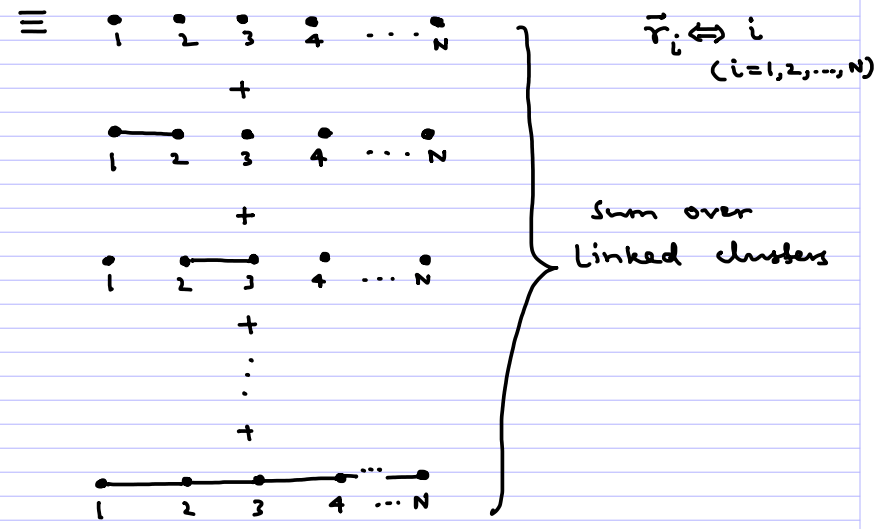
- low densities.
- high temperatures.

Insights

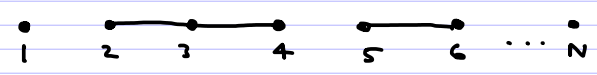
Consider the quantity C_N , s.t.,

$$C_N = \int \prod_{i=1}^N d^3 \vec{r}_i \left(1 + \sum_{i < j} f_{ij} + \sum_{\substack{i < j \\ k < l}} f_{ij} f_{kl} + \dots \right)$$

perturbation series



where a typical linked cluster



corresponds to the product

$$\left(\int d^3 r_1 \right) \left(\int d^3 r_2 d^3 r_3 d^3 r_4 f_{23} f_{34} \right) \left(\int d^3 r_5 d^3 r_6 f_{56} \right) \dots \left(\int d^3 r_N \right)$$

The procedure involves enumerating all linked clusters followed by a sum over all contributions.